

Silver(I)-Catalyzed Dual Activation of Propargylic Alcohol and Aziridine/Azetidine: Triggering Ring-Opening and Endo-Selective Ring-Closing in a Cascade

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 $[Ag(COD)_2]PF_6$ catalyzes the reaction between propargyl alcohols and *N*-tosylaziridines/azetidines leading to a diverse range of *N*,*O*-heterocycles, namely, oxazines, oxazepines, and oxazocines via ring-opening and ring-closing in a cascade

In recent years tandem catalysis has attracted widespread attention from synthetic chemists who aim at constructing complex molecular architecture from simple building blocks.¹ Tandem catalysis involving a transition metal is often distinguished by (i) selective binding of multiple substrates across the active site(s), (ii) making and breaking of multiple bonds in a cascade, and (iii) the presence of more than one distinct catalytic event.² Therefore the success in tandem catalysis is interalia dependent on the degree of synchronization of the above events.³ A major ongoing program in our laboratory is in late transition metal catalysis, where the goal is to realize differential binding, dualactivation, and coupling between a π -system and a substrate having a hard-donor center.⁴ During the course of our studies we were attracted by the potential of coinage metals in general, and silver(I) in particular, to bind/activate π -systems such as an arene, alkene, or an alkyne on one hand and substrates bearing a heteroatom such as N, O, and S on the other.^{5,6} These features make the coinage metals trustworthy candidates in facilitating three important reactions, namely (i) the opening of a heterocyclic ring by an internal/external nucleophile, (ii) the nucleophilic addition across an alkene/alkyne by an external nucleophile, and (iii) internal nucleophilic attack across an alkene/alkyne leading to a ring closure (Figure 1). $^{7-9}$



FIGURE 1. Coinage metal (M) assisted binding and ring-opening events.

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Note that, when stitched in a 2-in-1 or 3-in-1 cascade, the above three reactions can potentially generate myriad opportunities in multicomponent synthesis.¹⁰ Keeping the above in view, we searched the literature on Ag(I)-catalysis where such cascade reactions ultimately resulted in a heterocycle. However, such examples are few.¹¹ Continuing on our success in propargylic activation¹² and aziridine ring-opening¹³ we are delighted to present here a Ag(I)-catalyzed tandem ring-opening and ring-closing involving propargylic alcohol and aziridine or azetidine resulting in the formation of 6-, 7-, and 8-membered *N*,*O*-heterocycles, namely oxazines, oxazepines, and oxazocines.¹⁴ The reaction is characterized by two main catalytic events operating in a cascade. In the first event the aziridine or azetidine ring is opened via nucleophilic attack by propargylic alcohol. In the second event the tethered *N*-nucleophile in the intermediate attacks the alkynic/allenic appendage leading to 100% endoselective ring-closing (Scheme 1).

SCHEME 1



The model reaction between cyclohexyl-*N*-tosylaziridine **1a** and propargyl alcohol **2a** in the presence of $[Ag(COD)_2]PF_6(2 \text{ mol }\%)$ in dichloroethane proceeded well at ambient temperature. At the consumption of **1a**, Cs₂CO₃ (1 equiv) was added and the reaction was continued at 80 °C leading to the formation of 1,4-oxazepine derivative **3** in 72% isolated yield (Scheme 2, Table 1).

SCHEME 2



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TABLE 1. [Ag(COD)₂]PF₆ Catalyzed Ring-Opening/Ring-Closing of Aziridines and Azetidines with Propargyl Alcohol⁴⁴



^{*a*}Conditions: *N*-tosylaziridine (0.5 mmol), propargyl alcohol (0.5 mmol), and $[Ag(COD)_2]PF_6$ (2 mol %), dry dichloroethane (2 mL) Cs₂CO₃ (1 equiv). ^{*b*}Isolated yield.

Control studies (Scheme 3 and Supporting Information Scheme S1 and Table S1)¹⁵ indicated that (i) the optimum catalyst loading is 2 mol %; (ii) the reaction proceeds with AgPF₆, AgSbF₆, and AgBF₄ as the catalyst albeit with poorer yield of the product; (iii) other silver salts such as AgNO₃,

SCHEME 3



Ag₂CO₃, Ag₂SO₄, and AgCl do not promote any reaction; (iv) the product of the first catalytic event, namely 4-methyl-N-(2-propargyloxycyclohexyl)benzene sulfonamide **3'**, was isolated in 80% yield simply by quenching the reaction after 1 h; (v) the dual combination of [Ag(COD)₂]PF₆ and Cs₂CO₃ efficiently promoted the ring-closing of **3'** to **3**; and (vi) Cs₂CO₃ alone cannot promote the formation **3** and **3'** (Scheme 3). All of the above observations highlight the unique role of Ag(I) in this novel cascade reaction.

It is worth mentioning here that while this work was nearing completion, two reports on coupling between arylpropargyl alcohols and aziridines have appeared. However, we would like to emphasize that in sharp contrast to our results, (a) stoichiometric *t*-BuOK promoted coupling gave rise to exclusive formation of endocyclic morpholines,^{16a} while (b) catalytic Yb(OTf)₃ promoted coupling yielded exclusively indene derivatives.^{16b}

The cascade coupling has been satisfactorily extended to various propargyl alcohols, and substituted N-tosyl aziridines/azetidines leading to the selective formation of a diverse range of N,O-heterocycles, namely 1,4-oxazepines, 1,5-oxazocines, and 1,4-oxazines (Scheme 2, Table 1). N-Tosylaziridines having an alkyl or aryl appendage smoothly reacted with propargyl alcohols with terminal acetylenic linker to provide the corresponding 1,4-oxazepines 3-10 as the exclusive product (entries 1-8). Note that oxazepine 6 is obtained as a cis-trans mixture (1:1) possibly due to the conformational equilibrium of the fused 5- to 7-member bicyclic ring system (entry 4). Interestingly the reaction could be tuned to give 1,4-oxazines 11–13 by employing either a propargyl alcohol having internal acetylenic linker (entries 9 and 10), or by increasing the steric crowding around the aziridine framework (entry 11).

SCHEME 4

As discussed later, the formation of oxazepines and oxazines above is the outcome of the ring-closing step. Gratifyingly, we could also employ *N*-tosylazetidines in the present cascade coupling reaction to provide the corresponding 1,5-oxazocine derivatives **14**–**16** in moderate yields (entries 12-14).

While mechanistic details must await further studies, a proposal on the reaction course is presented in Scheme 4. The initial event will involve activation of the heteroatoms of the substrates by $[Ag(COD)]^+$.¹⁷ This will facilitate the opening of the aziridine/azetidine ring. π -Activation and metallotropic rearrangement of 17 is expected to result in the formation of intermediates Ag(I)-alkyne 17a, and Ag(I)-allene 17b. When assisted by a base, the ring-closing event from 17a and 17b may follow endo-dig (path a, path c) or exo-dig/trig (path b, path d) pathways-subsequent protodemetalation would afford the organic product and regenerate the Ag(I) catalyst. It is point worthy that ring-closure via path a involving 7-endo-dig or 8-endo-dig mode and path c involving 6-endodig mode are favored by Baldwin's rule. Indeed the observed formation of 1,4-oxazepines, 1,5-oxazocines, and 1,4-oxazines in our hand conforms to the above selection rules. Path b involving ring-closure in a 6-exo-dig manner is disfavored. Although the 5-exo-trig mode (path d) for the plausible formation of oxazolidine 25 is favored, we could not observe its formation.

In conclusion, we have presented a novel approach for the preparation of 6-, 7-, and 8-member unsaturated heterocycles having two heteroatoms through a Ag(I)-catalyzed cascade reaction of aziridines/azetidines with propargyl alcohols. Further work is underway in our laboratory to broaden the scope of the reaction, and to explore the mechanism.

Experimental Section

All reactions were carried out under an argon atmosphere in flame-dried glassware with Schlenk techniques. Chromatographic purifications were done with either 60-120 or 100-200 mesh silica gel. For reaction monitoring, precoated silica gel 60 F₂₅₄ TLC sheets were used. Petroleum ether refers to the fraction boiling in the range 60-80 °C. Dichloroethane was dried and distilled prior to use.

Preparation of $[Ag(COD)_2]PF_6$. To a water solution of AgPF₆ (0.5 mmol in 3 mL of water) was added a methanolic solution of



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1,5-cyclooctadiene (2 mmol in 0.5 mL of methanol) at room temperature in the absence of light. Reaction was completed just after 5 min and the residue was filtered and washed with water and hexane. The white solid was then dried under vacuum. It was then crystallized from benzene/ dichloromethane solvent system. Anal. Calcd for $C_{16}H_{26}OAgPF_6$: C, 39.44; H, 5.38. Found: C, 39.46; H, 5.04. MS (m/z) [Ag(C_8H_{12})]⁺ at 323 at cone voltage 15 V, [Ag(C_8H_{12})]⁺ at 215 at cone voltage 35 V.

General Procedure. A mixture of *N*-tosylaziridine (0.5 mmol), propargyl alcohol (0.5 mmol), and $[Ag(COD)_2]PF_6$ (2 mol %) in 2 mL of dry dichloroethane was stirred at room temperature. After 1–3 h (vide TLC), when aziridine was nearly diminished then Cs₂CO₃ (1 equiv) was added and the reaction mixture was heated at 80 °C for 3–5 h. After completion of the reaction (vide TLC) the mixture was diluted with water (5 mL) and extracted with dichloromethane. The combined organic layers were dried over anhydrous sodium sulfate and concentrated in vacuum, then the resulting product was purified by column chromatography on silica gel (100–200 mesh, ethyl acetate–petroleum ether, 1:5 v/v) to afford the corresponding heterocycles in good yields.

Preparation of 5-Tosyl-2,5,5a,6,7,8,9,9a-octahydrobenzo[b]-[1,4]oxazepine [3]. A mixture of N-tosylcyclohexyl aziridine (0.5 mmol, 125 mg), propargyl alcohol (0.5 mmol, 30 μ L), and [Ag(COD)₂]PF₆ (0.01 mmol, 4.6 mg) in 2 mL of dichloroethane was stirred at room temperature. After 1 h when aziridine was totally diminished Cs₂CO₃ (0.5 mmol, 163 mg) was added and the reaction mixture was heated at 80 °C for 3 h. After completion of the reaction (vide TLC), the reaction mixture was diluted with water (5 mL) and extracted with dichloromethane. The combined organic layers were dried over anhydrous sodium sulfate and concentrated in vacuum, then the resulting product was purified by column chromatography on silica gel (100-200)mesh, ethyl acetate-petroleum ether, 1:5). The yield was obtained in 72% (110.5 mg). ¹H NMR (400 MHz, CDCl₃) δ 1.31-1.46 (m, 4H), 1.68-1.75 (m, 2H), 1.92 (t, J = 5.2 Hz, 1H), 2.28-2.32 (m, 1H), 2.39 (s, 3H), 3.20-3.26 (m, 1H), 3.60-3.66 (m, 1H), 3.89-4.02 (m, 2H), 4.98-5.01 (m, 1H), 6.28 (d, J = 9.2 Hz, 1H), 7.25 (d, J = 8 Hz, 2H), 7.71 (d, $J = 10^{-10}$ 8.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 21.6, 24.8, 29.7, 33.3, 34.6, 67.9, 69.2, 78.4, 120.5, 124.9, 127.2, 129.3, 136.3, 143.2; DEPT-135 NMR 21.6 (CH₃), 24.8 (CH₂), 29.7 (CH₂), 33.2 (CH₂), 34.6 (CH₂), 67.9 (CH), 69.2 (CH₂), 78.3 (CH), 120.6 (CH), 124.8 (CH), 127.2 (CH), 129.4 (CH). Anal. Calcd for C₁₆H₂₁NO₃S: C, 62.51; H, 6.89; N, 4.56. Found: C, 62.49; H, 6.92; N, 4.54.

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Supporting Information Available: Experimental procedures, characterization data, and ¹H and ¹³C NMR spectra for all new compounds and crystallographic data for **4**, **9**, and **12**. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽¹⁵⁾ See the Supporting Information for details.

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⁽¹⁷⁾ ESI-MS spectrum of $[Ag(COD)_2]PF_6$ in ES(+) mode showed peaks corresponding to both $[Ag(COD)_2]^+$ and $[Ag(COD)]^+$ at low cone voltage (15 V). On the other hand at high cone voltage (35 V), $[Ag(COD)_2]^+$ unst de only species present. This suggests the lability of $[Ag(COD)_2]^+$ in solution. Note that the metal-olefin bond distance in $[Ag(COD)_2]PF_6$ (2.50 Å) is longer compared to other d¹⁰ analogues such as Cu, Ni, and Pt; see: (a) Vander, J.; Hende, H.; Baird, W. C. J. Am. Chem. Soc. **1963**, 85, 1003. (b) Schunn, R. A.; Ittel, S. D.; Cushing, M. A. Inorg. Synth. **1990**, 28, 94. (c) Howard, J. A. K. Acta Crystallogr. **1982**, 338, 2896.